



Knowledge that will change your world

## The Chemistry of the metabolome

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### What is a component of the metabolome?

- In the context of metabolomics, it is *compound of any origin that has a molecular weight <1,500 Da that can be detected in the biological system being studied*
- This is an arbitrary definition

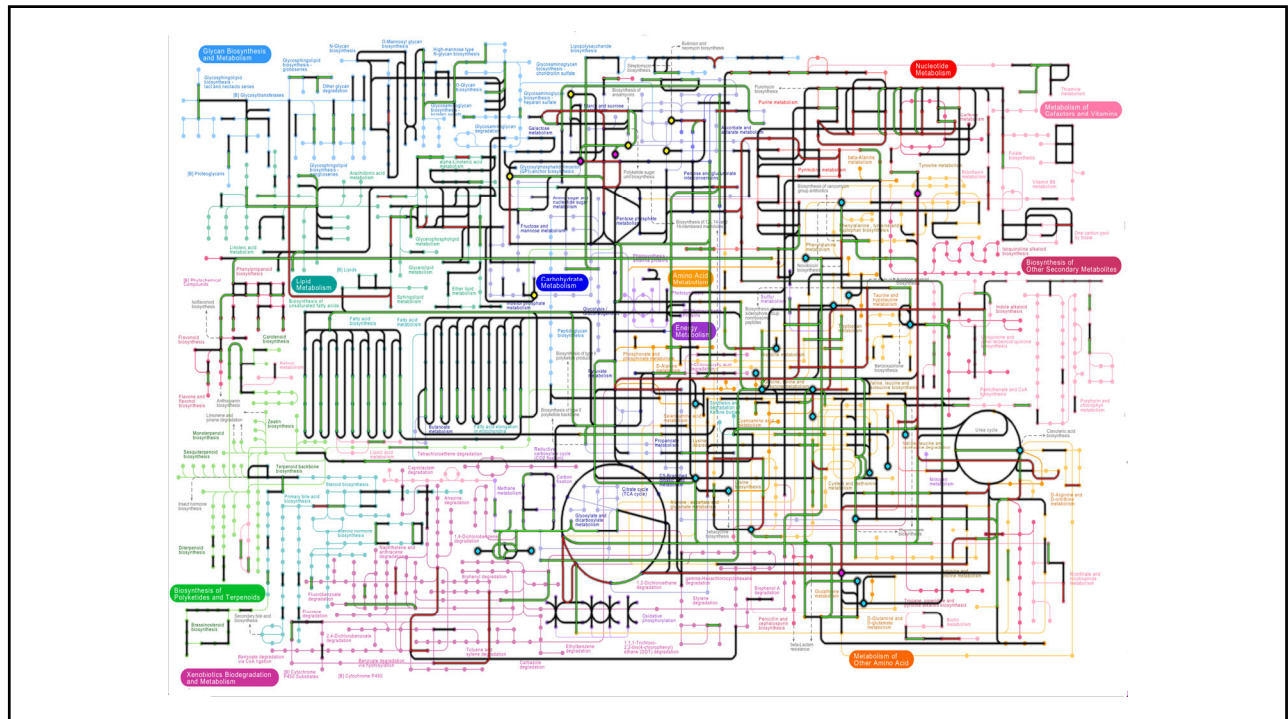
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## The metabolome is more than what's in textbooks

Metabolites synthesized from small molecule precursors by human cells

Metabolite pool in tissues and biofluids

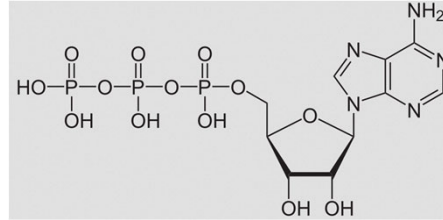
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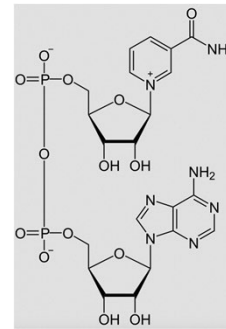
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## Critical metabolites

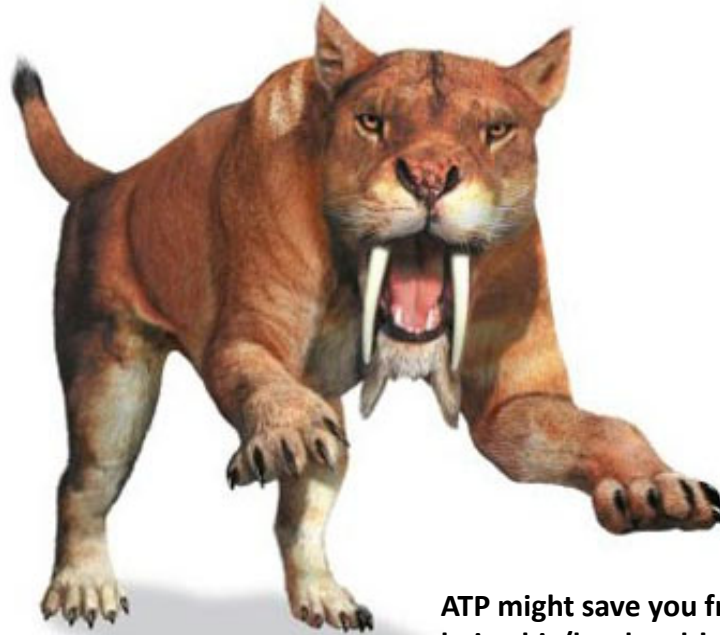
**ATP: adenosine-5-triphosphate**



**NAD<sup>+</sup>/NADH: nicotinamide adenine dinucleotide**

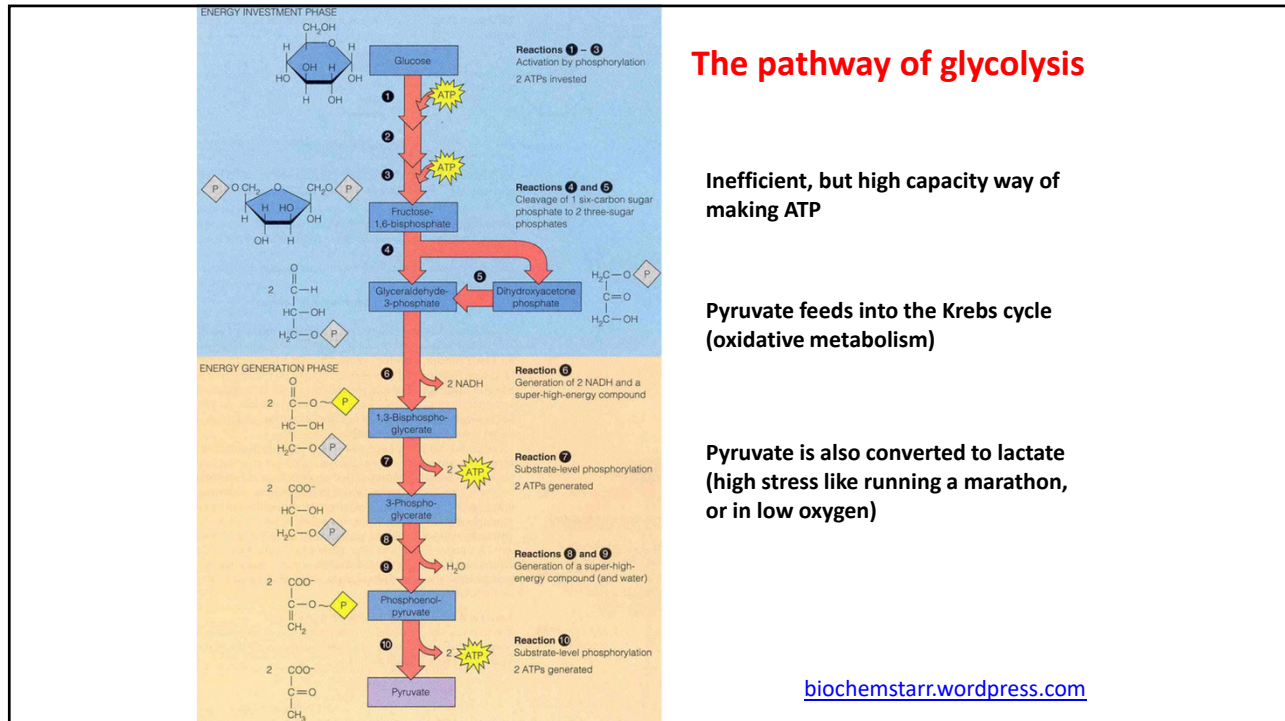


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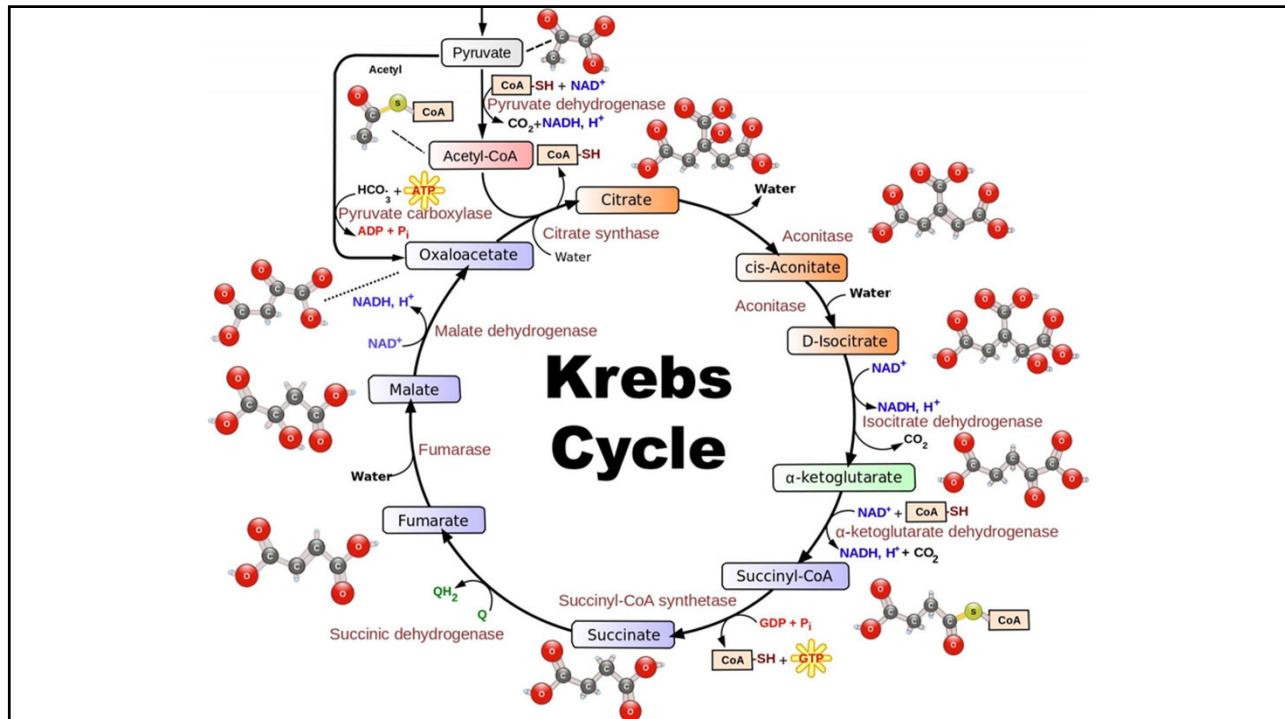


**ATP might save you from  
being his/her lunch!**

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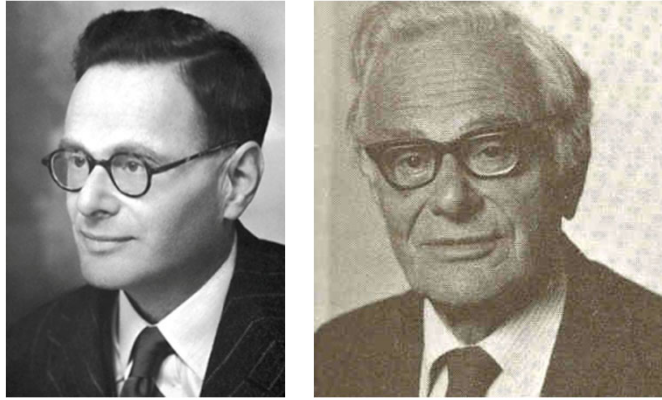


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## Sir Hans Krebs



Had the pleasure as a graduate student of introducing him at a seminar

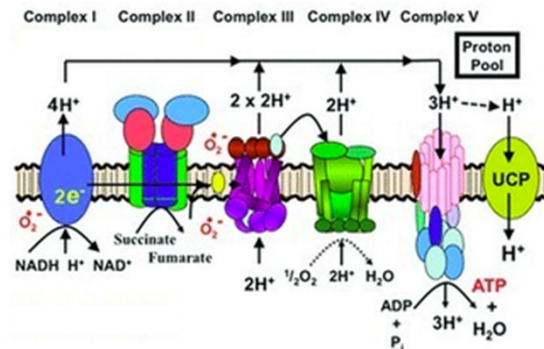
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## (Sir) Hans Krebs

- There was a young lady from Hyde
- Who ate a green apple and died
- Inside she lamented, the apple fermented
- And made cider inside her inside

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## Mitochondrial oxidative phosphorylation



NADH from the Krebs cycle, as well as succinate, generate a proton ( $H^+$ ) gradient (upper region) that drives rotation of one of the subunits of ATP synthase. This exposes the catalytic domain of this enzyme and makes ATP.

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## ATP synthetase

<http://www.mrc-mbu.cam.ac.uk/projects/2248/molecular-animations-atp-synthase>

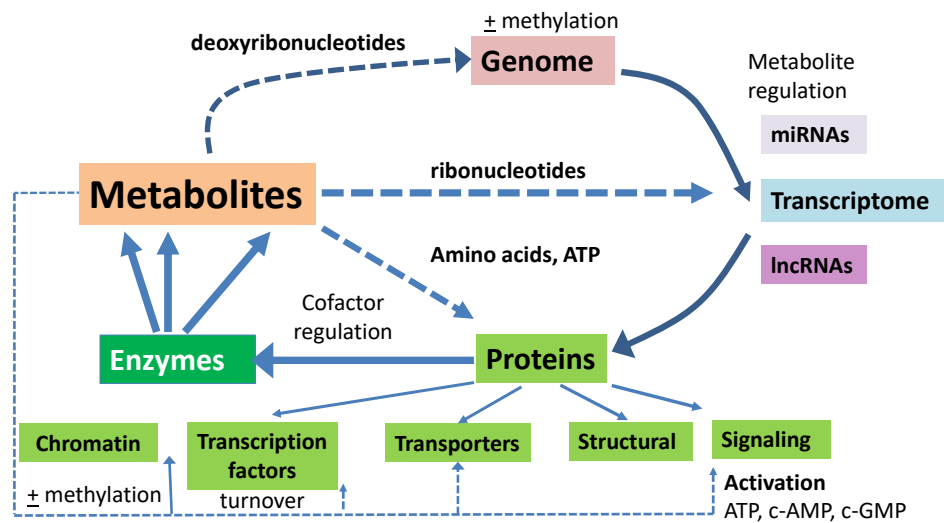
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## Understanding metabolites

- Metabolites represent the *action items* that come from gene expression and protein activity
  - They are found in the same range of concentrations as drugs
  - Metabolites ( $\mu\text{M}$  or  $\text{mM}$ ) (acetyl CoA, ATP, S-AdMet,  $\alpha\text{KG}$ ) are regulators of epigenetics
  - Bile acids ( $\mu\text{M}$ ) are the natural ligands of FXR and LXR
  - Other metabolites ( $\text{pM}$  or  $\text{nM}$ ) may be exquisite physiological regulators of kidney function (prostaglandins,  $\text{F}_2$ -isoprostanes)
- Studying the metabolome requires multiple levels of science from the analytical to the physiologic to the computational

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### Metabolites are associated with every aspect of cellular events



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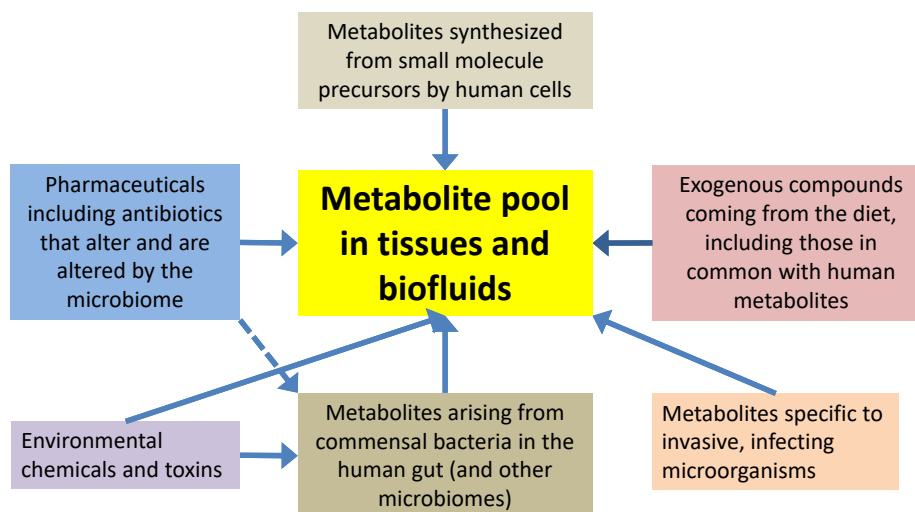
## Metabolism and time

- Not only should metabolites appear in the right place, there is also the question of the importance of the timescale
- Metabolism defects in the heart may be only seconds away from death – rogue waves in metabolism??
- Irreversible damage to the brain may occur in minutes
- Go/No-Go decisions for a cell to divide or apoptose may occur in tens of mins



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## The metabolome is more than what's in textbooks




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### Amino acids

Essential	Non-essential
Arg*	Ala
His	Asn
Ile	Asp
Leu	Cys
Lys	Gln
Met	Glu
Phe	Gly
Thr	Pro
Trp	Ser
Val	Tyr




Meat eater

Berry eater




Have to eat foods rich in these

↙

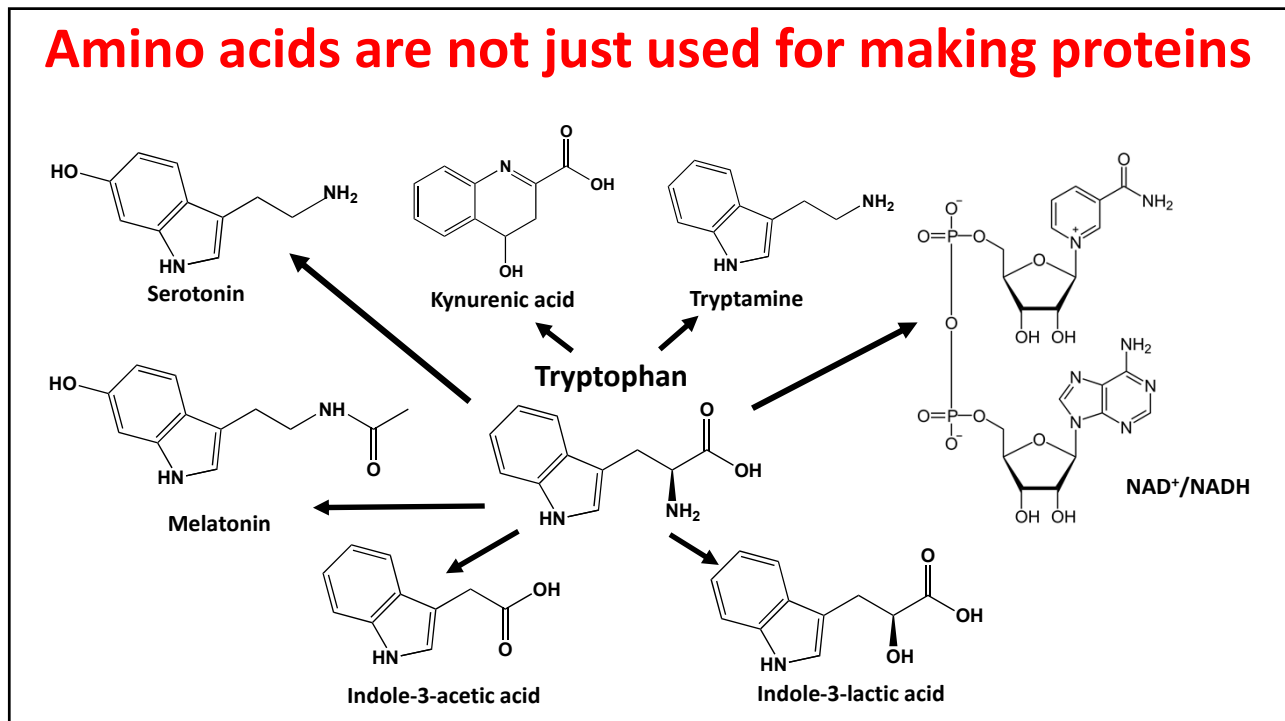


Vegetarian

Fruitarian

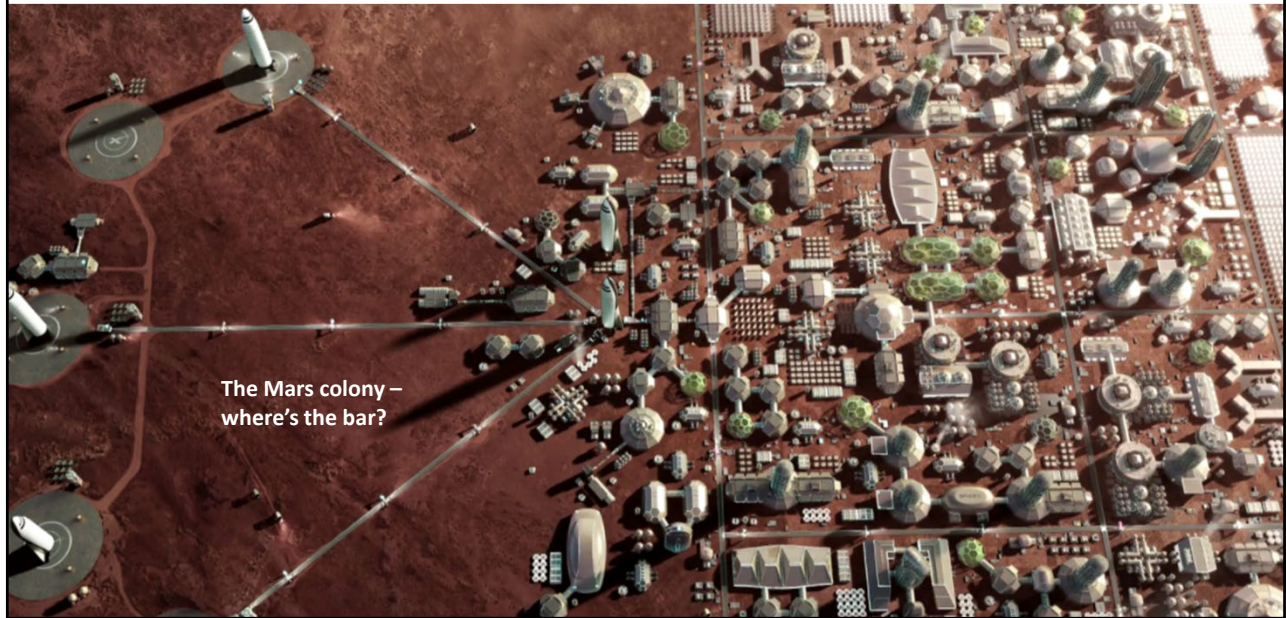


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## Important points to understand

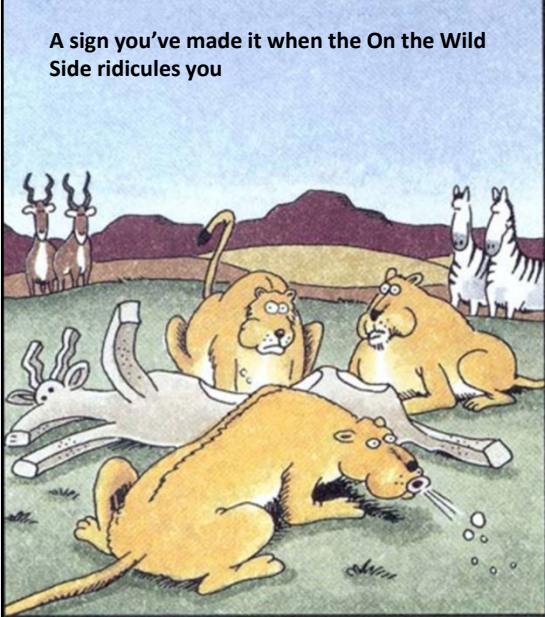


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Dark beer – B vitamins	Rich in polyphenols	Yoghurt +polyphenols	Tempeh - soy
			
Bitter beer – male hop UK	Alcohol/calories	Plain, but fermented	Kimchi - cabbage
			
			Ogi + hibiscus flower syrup

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**A sign you've made it when the On the Wild Side ridicules you**




In sudden disgust, the three lionesses realized they had killed a tofudebeest—one of the Serengeti's obnoxious health antelopes.

### Be kind to your "cat"

Vet. Pathol. 25:48-57 (1988)

#### Veno-occlusive Disease of the Liver in Captive Cheetah

*The main hepatic lesion was seen in 60% of the sexually mature cheetah (out of 126 captive animals). Observed in 1 year olds, but got worse with age and led to liver failure. Came from supplementation of the horsemeat diet with soy protein and the phytoestrogens therein.*



**Cats are exquisitely sensitive to aspirin and tylenol**

- The defect is in UGT1A6 which has become a pseudogene – the WT form glucuronidates phenols (a mechanism to excrete them)
  - Cats are hypercarnivores
    - Not exposed to modern drugs or plants in which there are substantial amounts of phenols
    - Victims of "Use it or lose it"
    - Diet-driven evolution
- Mutations in exon 1
  - Stop codons at bp 274-276 and 379-381 (>10 MYA)
- UGT1A1 that glucuronidates bilirubin is unaffected

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## Overview of metabolome chemistry

Metabolites encompass an enormous range of chemistries

- **Gaseous**
  - H<sub>2</sub>, H<sub>2</sub>S
- **Volatile**
  - Butyric acid, acetone, skatole
- **Hydrophilic (water-loving)**
  - Glucose
- **Charged-positive/negative**
  - Amino acids, nucleotides, organic acids, amines
- **Hydrophobic (fat-loving)**
  - Lipids, steroids, hydrocarbons

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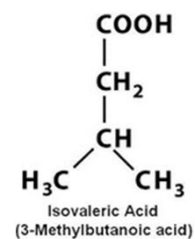
## Gases and volatiles

- **In breath**
  - H<sub>2</sub> from reductive anerobic bacteria
    - Lactose-intolerant
    - Measure of gut transit (typically 4-6 hours)
  - CO<sub>2</sub>
    - From all carbon-containing substrates
    - From specific <sup>13</sup>C-labelled substrates
  - Acetone (in diabetics)
  - Trimethylamine
    - From fish, or flavin monooxygenase (FMO3)-deficient subjects

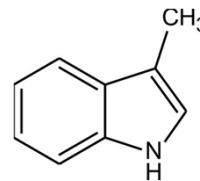
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## Gases and volatiles

- **Sweat gland**
  - Sweaty socks syndrome
    - **Isovaleric acid** (leucine metabolism)
    - Caused by bacteria or enzyme defect



- **Flatulence**
  - Mostly gases (H<sub>2</sub>, CO<sub>2</sub> and H<sub>2</sub>S), but with volatiles produced by colonic bacteria (**skatole**, from the amino acid tryptophan)



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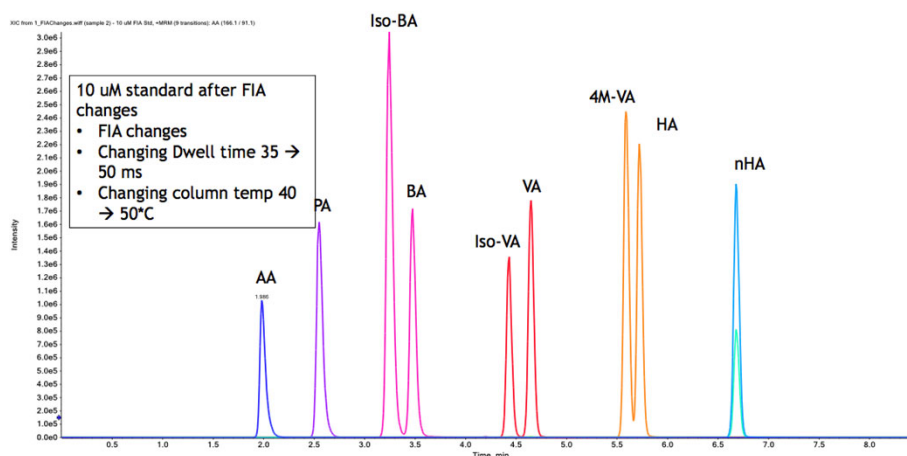


## Other volatiles

- Short chain, unsubstituted fatty acids
  - Formic, acetic, propionic, butyric acids
- Will evaporate in the acidic form
  - Formic acid, b.p. 101°C
  - Acetic acid, b.p. 118°C
  - Propionic acid, b.p. 141°C
  - Butyric acid, b.p. 163.8°C
  - Isobutyric acid, b.p. 155°C
- React *in situ* to form a non-volatile derivative before evaporating

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## *o*-benzylhydroxylamine derivatives

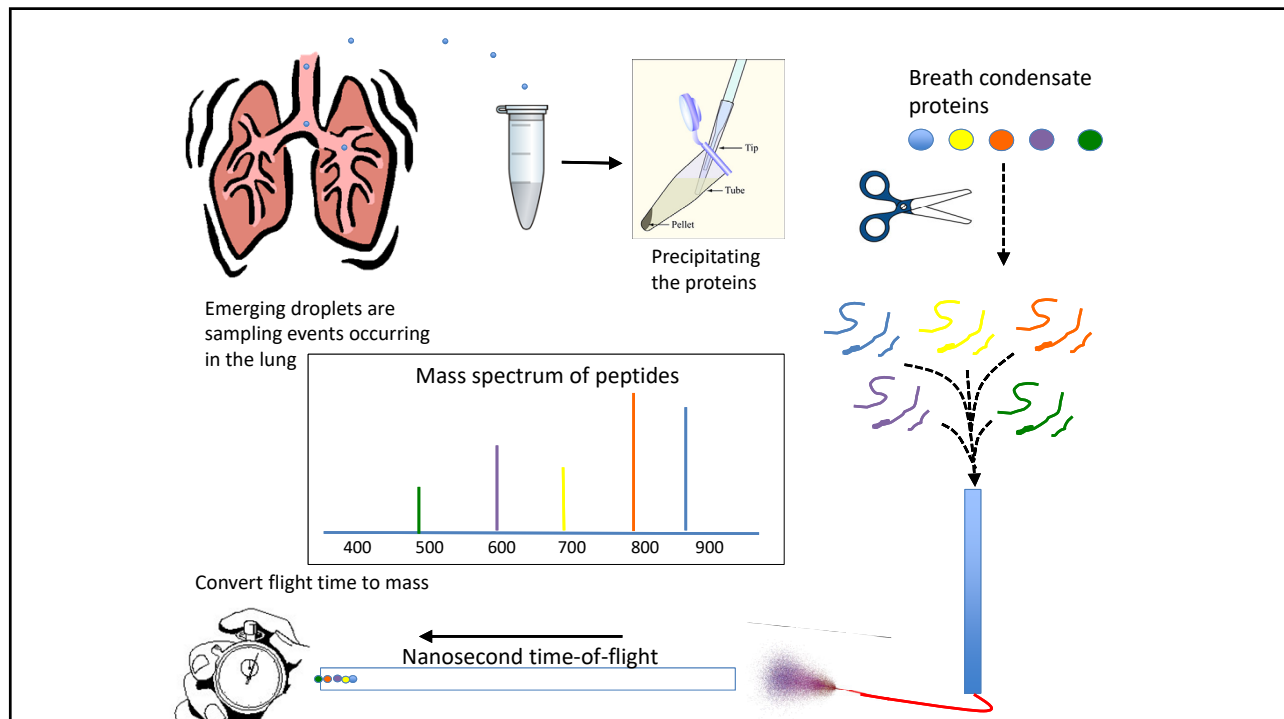


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## Breath condensates

- Not strictly consisting of volatiles
- A mist or spray created by the frothing of the fluids inside the lung
  - Condensable using a dry-ice cooled trap
  - Several ml of condensate can be easily collected in 5-8 min

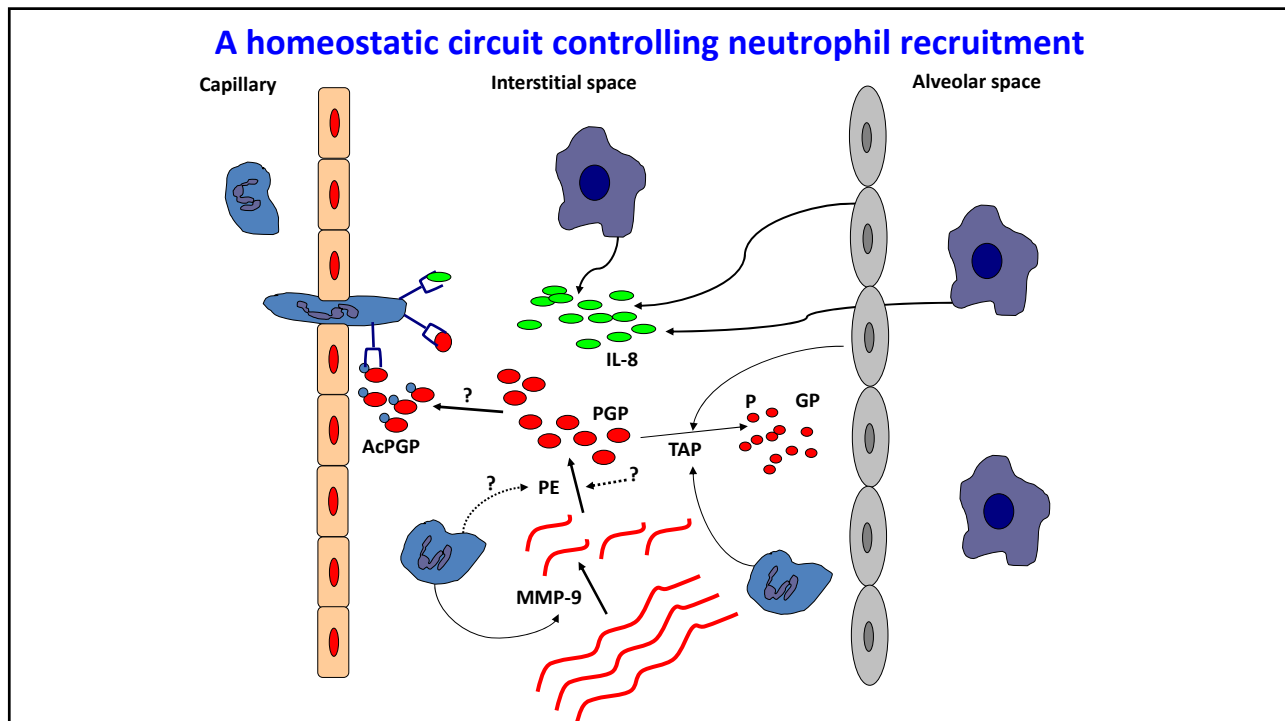
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## Certain metabolites are peptides

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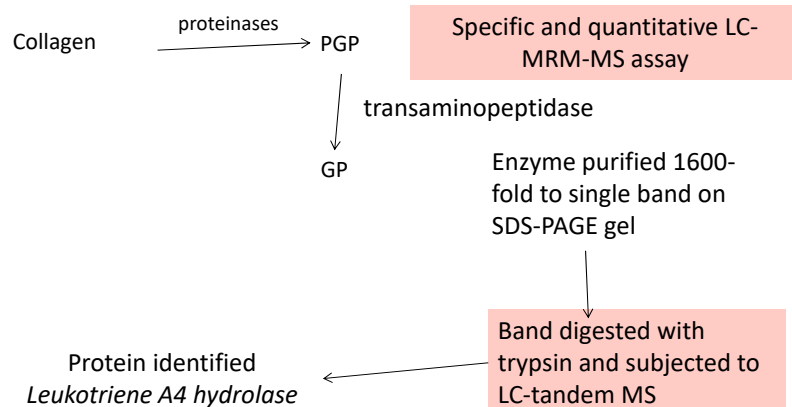
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## PGP is a common peptide in human collagen

MFSFVDLRLLLLLLAATALLTHGQEEGQVEGQDEDIPPIITCVQNGRLRYHDRVWKPPEPCR I  
 CVCDNGKVLCDDDVICDETKNCPGAEVPEGECCPVC PDGSESPDQETTGVVEGPKGDTGPR  
 GPRGPAGPPGRDGI PGQPLPGP**PGPPGPPGP**PGLGGNFAPQLSYGYDEKSTGGISV**PGP**  
 MGPSGPRGL**PGP**PGA**PGP**QGFQGPPEPEPEPGASGPMGPRGP**PGP**PGKNGDDGEAGKPR  
 PGERGP**PGP**QGARGLPGTAGLPGMKGHRGFSGLDGAKGADAGPAGPKGEPGSPGENGAPGQ  
 MGPRGLPGERGRPGA**PGP**AGARGNDGATGAAGP**PGP**TGPAGPPFPFPGAVGAKGEAGPQGP  
 RGSEGPQGVVRGE**PGPPGP**AGAAGPAGNPGADGQPGAANGAPGIAGAPFPARGPSPG  
 QGPGGP**PGP**KGNSGEPGAPGSKGDTGAKGE**PGP**VGVQGP**PGP**AGEEGKRARGE**PGP**TGL  
**PGP**PGERGGPSRGPAGDGVAGPKGPAGERGS**PGP**AGPKGSPGEAGRPGEAGLPGAKGL  
 TGSPGS**PGP**DGKGTGP**PGP**AGQDGR**PGPPGP**PGARQAGVMGF**PGP**KGAAGEPGKAGERV  
**PGP**PGAVGPAGKDEAGAQQFP**PGP**AGPAGERGEQQFAGSPGFQGL**PGP**AGFPGEAGKPGE  
 QGVVPGDLGA**PGP**SGARGERGFPGERGVQGP**PGP**AGPRGANGAPGNDGAKGDAGAPGAPGS  
 QGAPGLQGMPPGERGAAGL**PGP**KGDRGDAGPKGADGSPKDGVRGLTGP I GPPGPAGAPGD  
 KGESGPSGAPPTGARGAPGDRGE**PGPPGP**AGFAGPPGADGQPGAKEPEGDAGAKGDAGP  
**PGP**AGPAGP**PGP**IGNVGPAGAKGARGASAGPPGATGFPGAAGRVP**PGP**SGNAGP**PGPPGP**  
 AGKEGKGRGETGPAGRPGEVGP**PGPPGP**AGEKSGADGPAGAPGT**PGP**QGIAGQRGV  
 VGLPGQRGERGFPL**PGP**SGEPGKQGPSGASGERGP**PGP**MGPPGLAGPPGESGREGAPGA  
 EGSGRDGS PGAKDRGETGPAGPPGAPGAPGA**PGP**VGPAGKSGDRGETGPAGPAGVGP  
 VGARGPAGPQGPRGDKGETGEQDRIKHRGFSGLQGP**PGP**PGSPGEQGPSGASGPAGP  
 RGPPGSAGAPGKDLNGL**PGP**IGP**PGP**RGRTGDAGPVGP**PGPPGPPGPPGPP**PSAGFDFS  
 LPQPPQEKAHDDGRYYRADDANVVRDRDLEVDTTLKSLSQQIENIRSEPSRKNPARTCR  
 DLKMCHSDWKSGEYWIDPNQGCNLDAIKVFCNMETGETCVYPTQPSVAQKNWYISKNPKD  
 KRHVWFGESMTDGFQFEYGGQSDPADVAIQLTFLRLMSTEASQNIYHCKNVAYMDQQ  
 TGNLKKALLLQGSNEIEIRAEGNSRFTYSVTVDGCTSHTGAWGKTVIEYKTTKTSRLPI I  
 DVAPLDVGAPDQEFDFVGPVCF

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## Mass spec contribution to PGP story

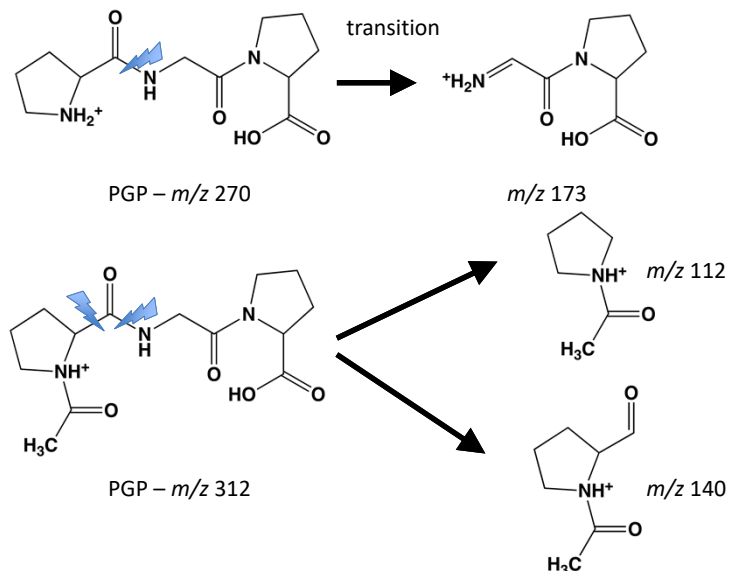


(Robert Snelgrove et al. Science, 2010)

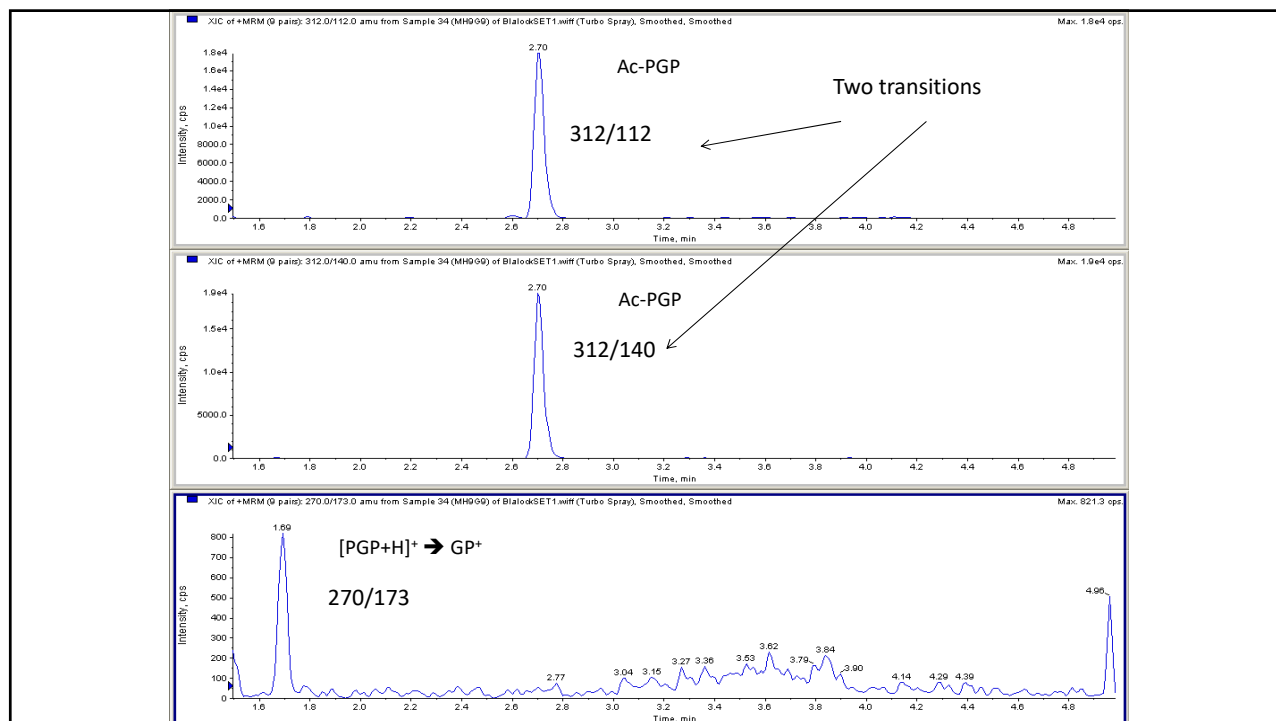
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## Measuring PGP and acetyl-PGP



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## **Metabolopeptidomics or peptidometabolomics**

- **Are peptides metabolites?**
- **Are the tripeptides real?  
Or is their mass simply coincident with the empirical formula of another metabolite?**

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## **Considering the case for tripeptides**

- **Examine the basic physiology and pharmacology**
- **Are there examples of bioactive tri-peptides?**
- **What about other oligopeptides?**
- **Where would they come from?**
- **Why does METLIN seem to always have tri- and not other oligopeptides?**

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**Tripeptides could come from foods,  
but are hydrolyzed by peptidases in  
the enterocyte to amino acids**

**Deficiencies in the peptidases could lead to  
food and bacterial peptides entering the  
systemic circulation**

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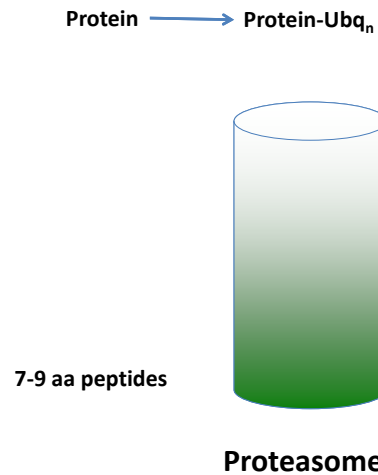
### **Can tripeptides have biological activity?**

- **For toxicologists, there is one very familiar tripeptide without whom, I would not be giving this talk, or you to listen to it.**
- **Glutathione (GSH) – glutamyl-cysteinyl-glycine**
  - GSH reacts with free radicals to generate GSH conjugates and therefore protects many organs
- **It is synthesized from small molecule precursors**
  - However, it is a true metabolite, i.e., it is made from smaller precursors without the direct aid of ribosomes

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## Are there other sources of short peptides?

- Proteins undergo degradation in the proteasome caused by targeted ubiquitination
  - The digested products are peptides (escapees?)
- Lysosomes
- Autophagosome
- Neutrophil attack
- Other proteases (in renal tubules?)
- Foreign antigens hydrolyzed and presented on surface of cells



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## Hydrophilic metabolites

- The most extreme hydrophilic metabolites without charged groups are the polyols:
  - Monosaccharides
    - Glucose
    - Fructose
  - Disaccharides
    - Lactose
    - Maltose
  - Oligosaccharides


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## Organic acids

- Besides the short chain fatty acids mentioned earlier, there are many organic acids representing important cellular pathways
  - Glycolytic intermediates
    - Glucose-1-P, Glucose-6-P, Fructose-6-P, Fructose-1,6-DP, Glyceraldehyde-3-P, Dihydroxyacetone-P, Glycerate-3-P, Phosphoenol-P, Pyruvate, Lactate
  - Krebs cycle
    - Citrate, cis-Aconitate, Iso-Citrate,  $\alpha$ -ketoglutarate, Succinate, Fumarate, Malate, Oxaloactate and those resulting from pathway defects
  - Nucleotides
    - ATP, ADP, AMP, GTP, etc.

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## How could we isolate organic acids?

- Organic acids at neutral pH are negatively charged
  - They will bind to anion exchange resins in say the formate form
- 

AG-1
- Can be eluted with ammonium formate acetate (mass spec compatible)

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## How could we isolate amino acids?

- Amino acids at neutral pH are positively charged
- They will bind to cation exchange resins in the H<sup>+</sup> form



AG-50

- Can be eluted with ammonium hydroxide (mass spec compatible)

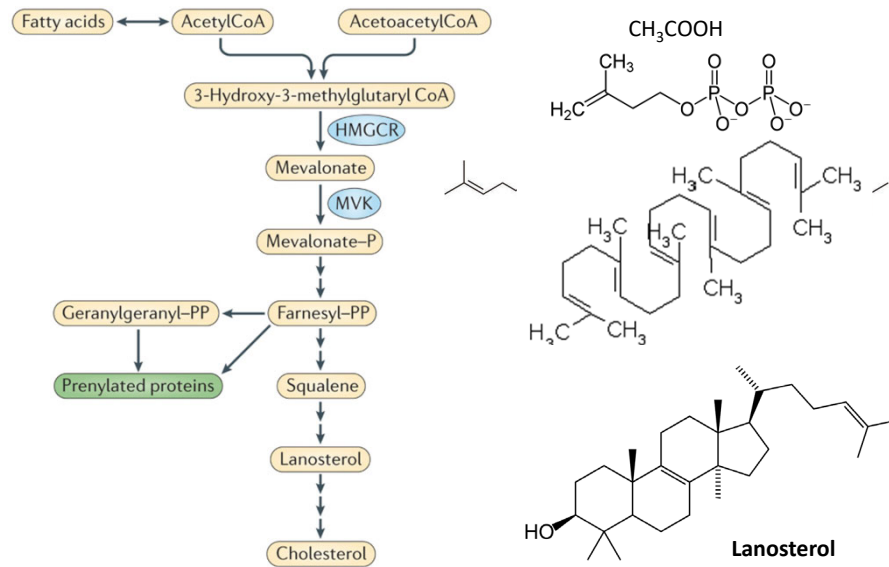
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## Hydrophobic metabolites

- These include sterols, steroid hormones, terpenoids, bile acids, vitamins A, D, E and K, and a vast array of lipids

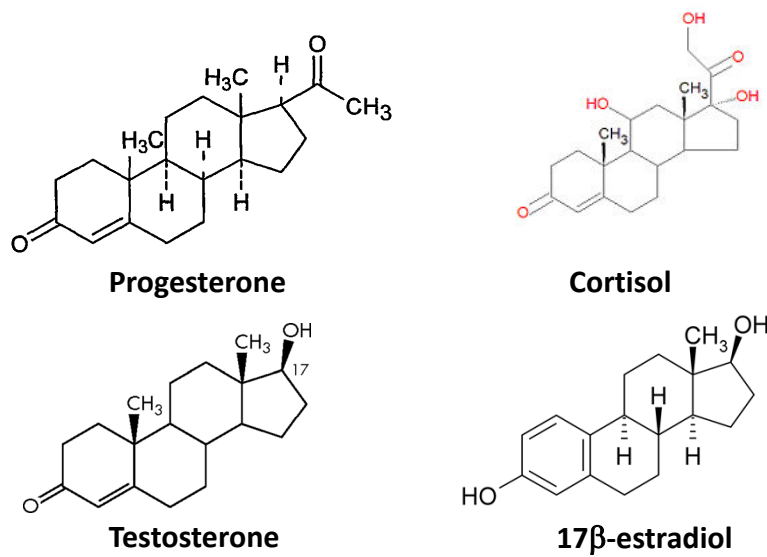
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## Isoprenoids and sterols



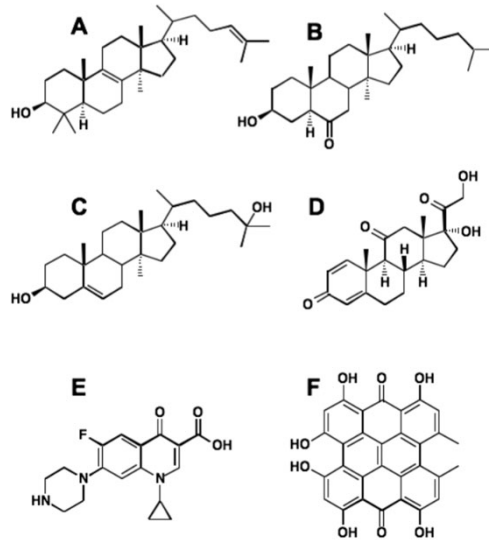
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## Steroids



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## Importance of sterols and other compounds in lens cataracts

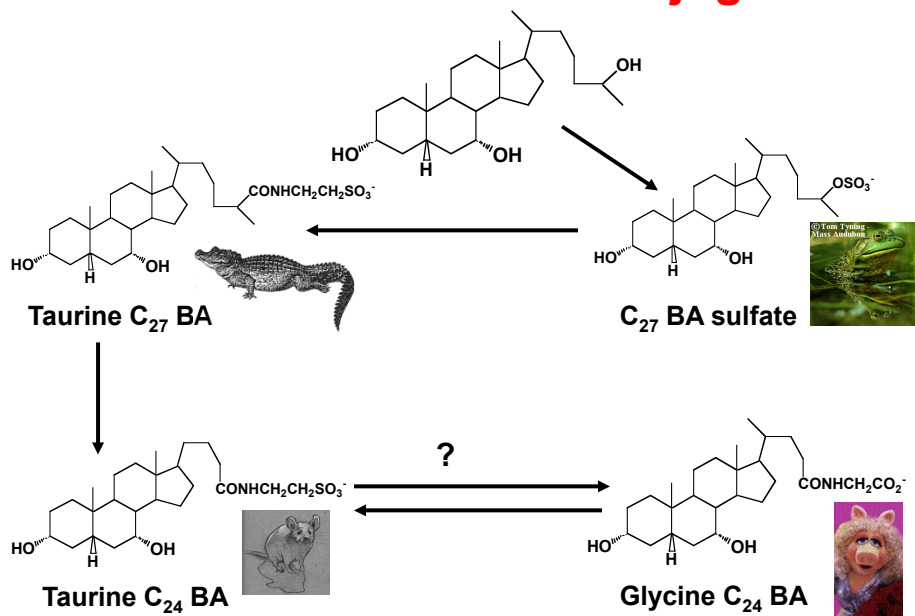


Structures A, B and C (all sterols) have recently been shown to have the property of “dissolving” lens cataracts. Cholesterol, on the other hand, has no effect. Other sterols observed in *cerebrotendinous xanthomatosis* promote cataracts.

D, E and F all promote lens cataracts. D is prednisone (an anti-inflammatory steroid), E is ciprofloxacin (an antibiotic) and F is hypericin from the botanical, St. John’s wort.

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## Evolution of bile acid conjugation



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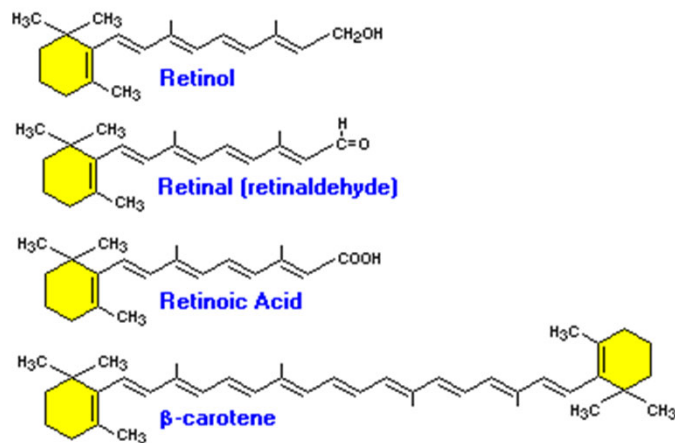


## The vitamins

Lack of these leads to serious illness, but not death

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## Vitamin A

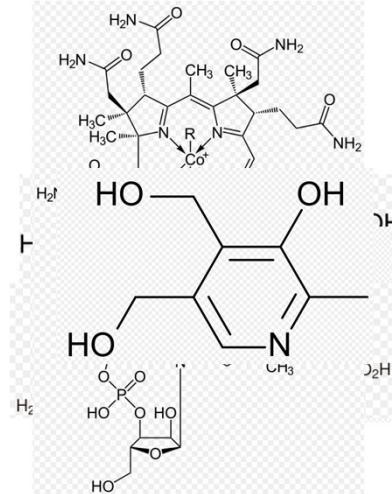


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## Vitamin B

- They are all water-soluble

- Vit B<sub>1</sub> – thiamine
- Vit B<sub>2</sub> – riboflavin
- Vit B<sub>3</sub> – niacin
- Vit B<sub>5</sub> - pantothenic acid
- Vit B<sub>6</sub> – pyridoxine
- Vit B<sub>7</sub> – biotin
- Vit B<sub>9</sub> – folic acid
- Vit B<sub>12</sub> – cobalamins

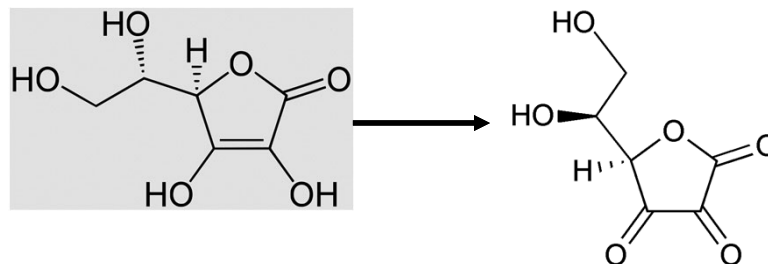


They are not made by human enzymes and if deficient in the diet cause disease

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## Vitamin C

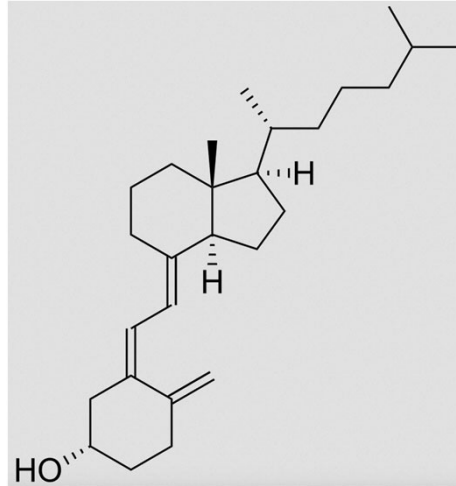
- Ascorbic acid



dehydroascorbic acid

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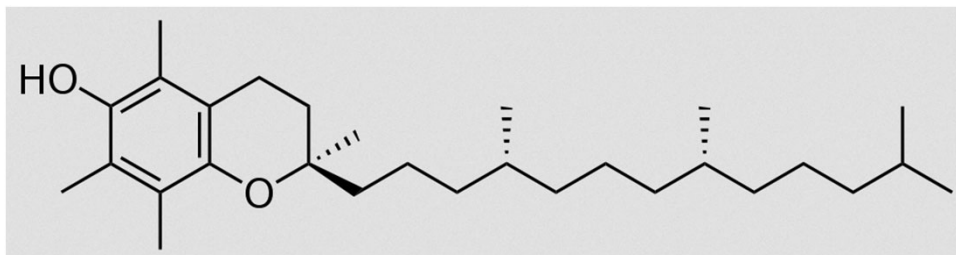
## Vitamin D



In fish, supplemented in milk, made in skin by UV light

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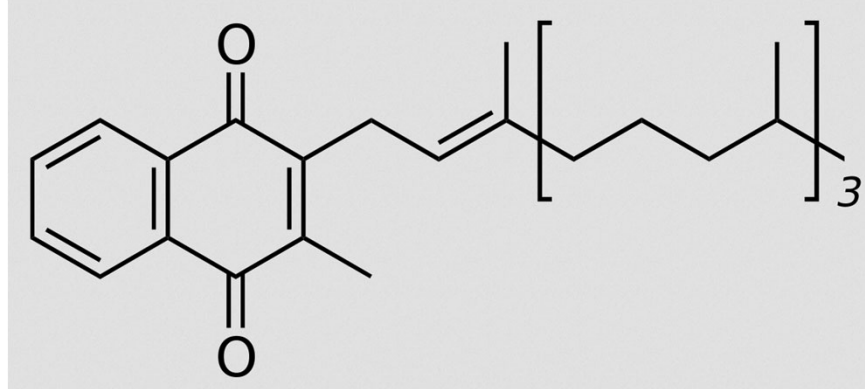
## Vitamin E



Found in oils from plants

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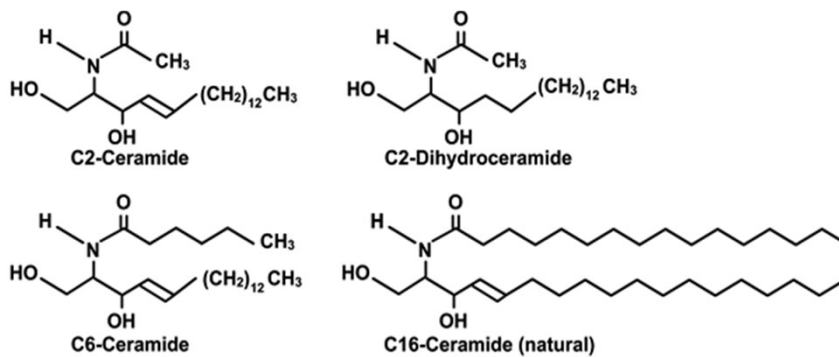
## Vitamin K



Is an anticoagulant – needed to stop bleeding

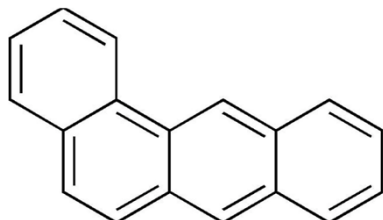
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## Phospholipids



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## Hydrocarbons



**Benz[a]anthracene**  
In smoke from barbecued meat



**Cetyl palmitate**  
In hair shampoo

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## Solubilities of the metabolites

- Those in biological fluids are “in solution”, but may not be soluble in water or methanol alone
  - Are glucose or amino acids soluble in methanol?
  - Are cholesterol esters in plasma soluble in methanol or water?
    - If a metabolite binding protein is precipitated by methanol, does the metabolite still bind to it?
  - Does pH have an effect on solubility?

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**Etc., etc.**